

The Oxygen Defect Perovskite $\text{Sr}_2\text{Mn}_2\text{O}_5$: HREM Study

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$\text{Sr}_2\text{Mn}_2\text{O}_5$ was investigated by high-resolution electron microscopy (HREM). Numerous crystals are characterized by a regular contrast, corresponding to the tunnel structure $\text{Sr}_2\text{Mn}_2\text{O}_5$, which can be described as an ordered oxygen-defect perovskite ($a = a_p \sqrt{2}$, $b = 2a_p \sqrt{2}$, and $c = a_p$). Several types of defects were observed and interpreted in terms of different orderings of oxygen vacancies: $\text{Sr}_2\text{Mn}_2\text{O}_5$ -type domains with antiphase boundaries (APB), $\text{Sr}_2\text{Mn}_2\text{O}_5$ -type domains with perovskite domains, new ordered domains ($a = a_p \sqrt{2}$ and $b = 4a_p \sqrt{2}$), and stacking faults. © 1986 Academic Press, Inc.

Introduction

It is well known that electrocatalytic and electrode properties of the oxygen-defects perovskites ABO_{3-x} have their origin in the nature of the *B* cations which can take simultaneously several oxidation states and several coordinations as shown, for example, for copper oxides (1). In this respect, manganese perovskites AMnO_{3-x} (2-10) seem to offer large possibilities due to the ability of manganese to take the two oxidation states, III and IV, and to have both octahedral and pyramidal coordinations. Besides the calcium perovskites whose structures have been recently studied (2-7), several phases have been isolated in the case of strontium manganese oxides (8, 9). However, these latter oxides, which exhibit complex arrangements of the oxygen vacancies have not been studied by electron microscopy so that little is known about their defect structure. In a recent paper

(10), it was shown that $\text{Sr}_2\text{Mn}_2\text{O}_5$ could be synthesized with the $\text{Ca}_2\text{Mn}_2\text{O}_5$ (7) structure. The present work deals with the high-resolution electron microscopy, HREM, study of the defects which appear in the crystals corresponding to samples with this nominal composition.

Experimental

The synthesis of $\text{Sr}_2\text{Mn}_2\text{O}_5$ has already been described (10). It consists in heating adequate mixtures of SrCO_3 and Mn_3O_4 first in air at 1150 and 1450°C successively, then in evacuated ampoule at 500°C in the presence of zirconium. The oxygen amount was determined by chemical analysis as previously described (10): only the samples corresponding to the composition $\text{Sr}_2\text{Mn}_2\text{O}_5$ were examined by electron microscopy.

High-resolution electron microscopic observations were made at 120 kV with a JEM 120 CX electron microscope equipped with

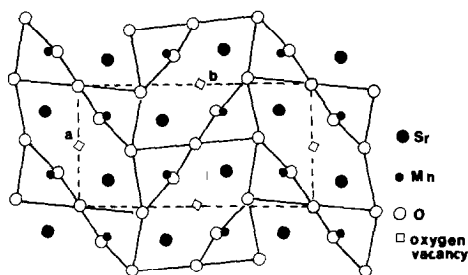


FIG. 1. Projection along [001] of the oxygen-defect perovskite $\text{Sr}_2\text{Mn}_2\text{O}_5$.

a top-entry double tilt ($\pm 10^\circ$) goniometer specimen holder and an objective lens with spherical (C_3) aberration coefficient of 0.7 mm. The samples were crushed in an agate mortar and then mounted on a holey carbon film. Thin fragments were oriented, mainly with [001] exactly parallel to the electron

beam, as indicated by the selected-area electron diffraction patterns. Image astigmatism was corrected by observing the granularity of the support carbon films and areas of thin oriented crystals, protruding over holes, were chosen for imaging. Multislice N-beam calculations were carried out with programs supplied by Skarnulis (13). Through-focus series were calculated for various thicknesses of the crystals.

Structure of $\text{Sr}_2\text{Mn}_2\text{O}_5$. The structure of $\text{Sr}_2\text{Mn}_2\text{O}_5$ (10) is characterized by an orthorhombic cell with $a = 5.523(1) \text{ \AA}$, $b = 10.761(5) \text{ \AA}$, and $c = 3.811(1) \text{ \AA}$. The host lattice “ Mn_2O_5 ” (Fig. 1) is built up from corner-sharing MnO_5 pyramids forming pseudo-hexagonal tunnels running along $\langle 001 \rangle$ and perovskite tunnels running

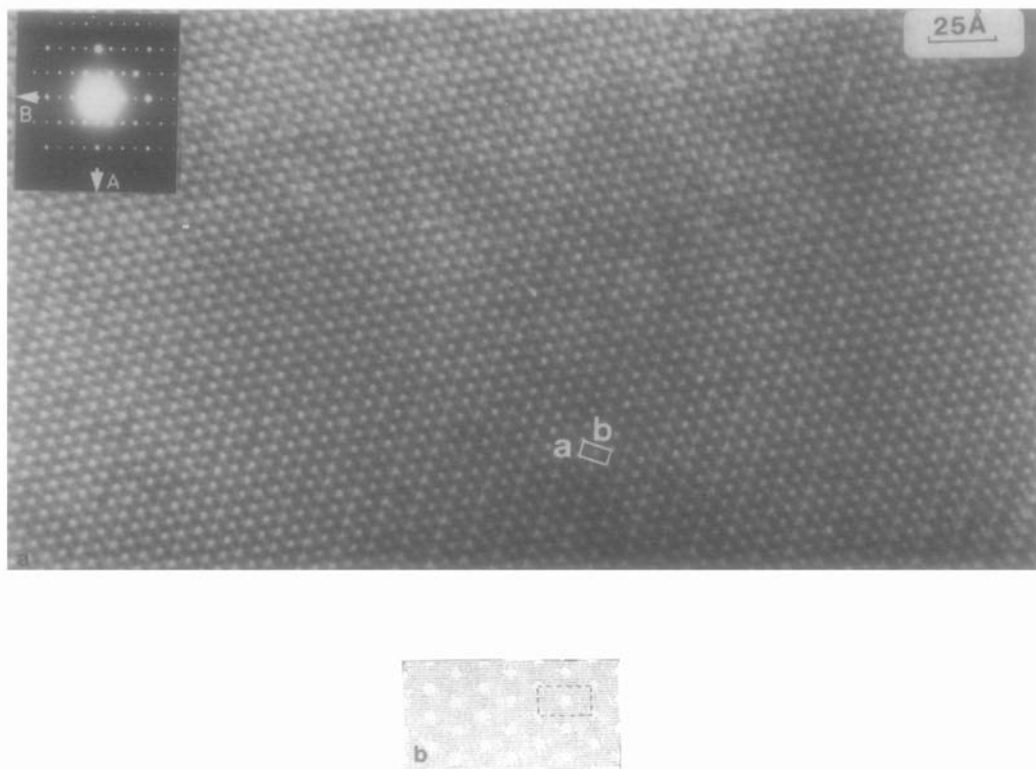


FIG. 2. (a) Lattice image of a crystal of $\text{Sr}_2\text{Mn}_2\text{O}_5$ obtained from [001] zone diffraction pattern. (b) Calculated image (4×4): thickness = 38 \AA , defocus setting $F = -600 \text{ \AA}$. (The drawn cell is translated from $a/2$ with regard to the crystallographic origin.)

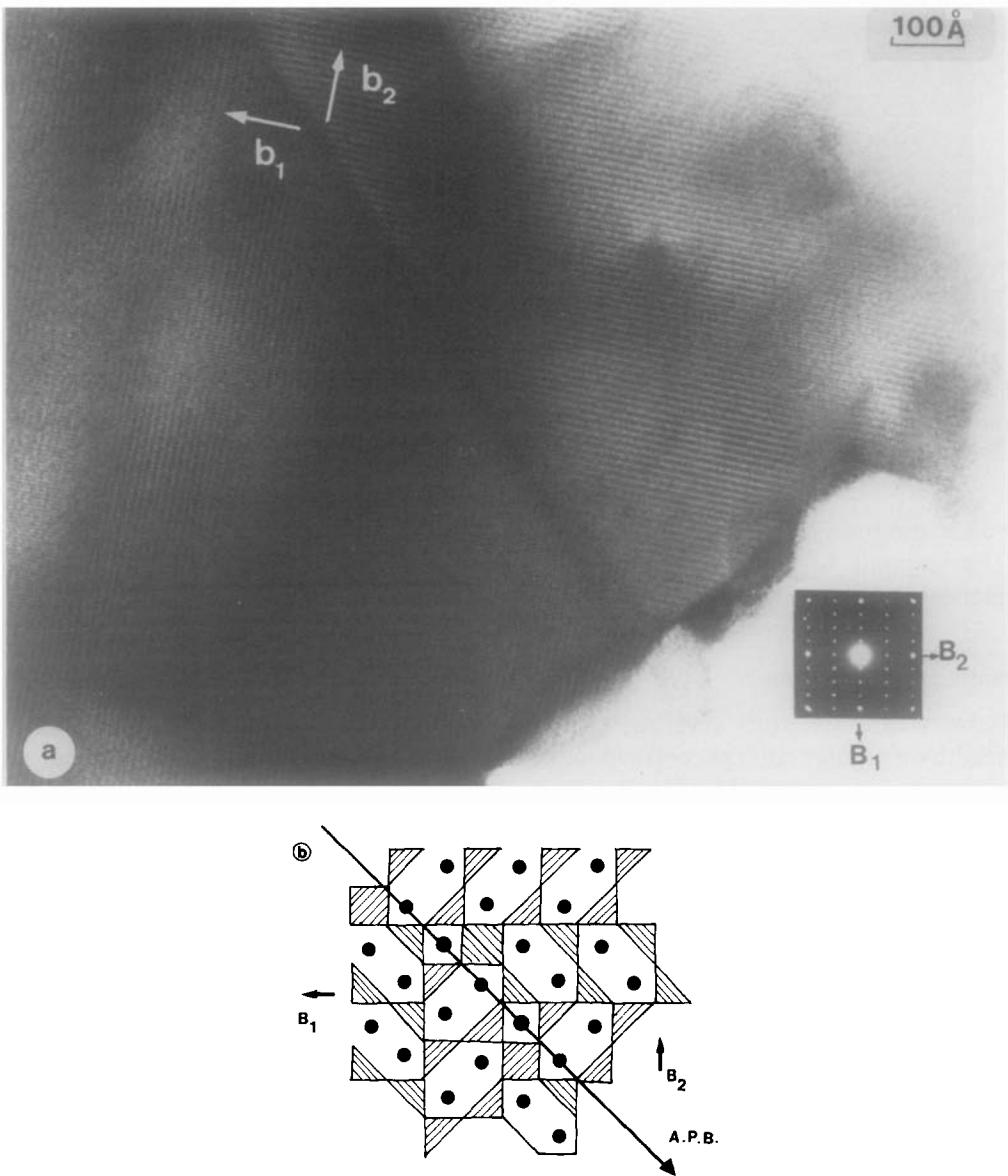


FIG. 3. (a) Micrograph, as a projection along $[001]$, of two adjacent microdomains exhibiting a $\text{Sr}_2\text{Mn}_2\text{O}_5$ -type structure. They are characterized by an angle of 90° between the two b axis: (b) Idealized drawing of the two orientated domains and their junction.

along $\langle 110 \rangle$ and $\langle \bar{1}\bar{1}0 \rangle$. The occurrence of oxygen-defect ordering results in a tilting of the MnO_5 pyramids around the c axis; the angles between the oxygen of two neighboring pyramids is close to 74° (instead of 90°

in the ideal structure). A large number of crystals exhibit images of this tunnel structure as shown from Fig. 2a taken along the $[001]$ zone. In this image it was concluded from the calculated images (Fig. 2b) that

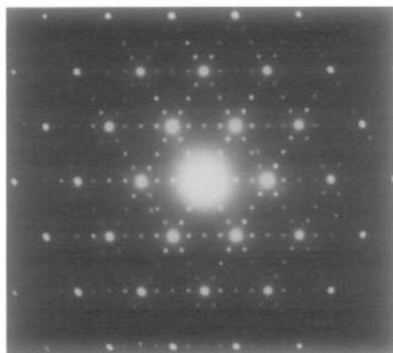


FIG. 4. Electron diffraction pattern showing the superstructure ($a \approx 2a_p \sqrt{2}$) along the three directions $\langle 110 \rangle_p$.

the alternate rows of bright dots, spaced 5.5 Å, correspond to the pseudo-hexagonal tunnels (10).

Results and Discussion

Besides the numerous crystals, characterized by a regular contrast corresponding to the tunnel structure of $\text{Sr}_2\text{Mn}_2\text{O}_5$, many crystals were found to exhibit defects. This work is focused on the main types of defects which were observed and the possible corresponding arrangement of the oxygen vacancies.

Sr₂Mn₂O₅-Type Domains and Antiphase Boundaries

The observations along \tilde{c} show that many crystals exhibit twinned domains exhibiting the $\text{Sr}_2\text{Mn}_2\text{O}_5$ structure. They are characterized (Fig. 3) by an angle of 90° between the two “ b ” directions and an orientation domain boundary parallel to (120), i.e., to $(100)_p$. The junction between two adjacent domains is easily explained by a slight local variation of stoichiometry (Fig. 3b) involving the presence of MnO_6 octahedra at the domain boundary. It is evident that such a variation of oxygen content could be too small to be detected by chemical analysis. This tendency to form do-

main comes from the fact that the ordering of the oxygen vacancies in the original cubic perovskite cell (a_p), leading to the superstructure $a \approx a_p \sqrt{2}$ and $b \approx 2\sqrt{2}a_p$, can take place along three equivalent directions in the perovskite matrix. Indeed many crystals exhibit an electron diffraction pattern of the (041) plane, i.e. of $(111)_p$, characterized by the expected superstructures along the three directions $\langle 110 \rangle_p$ with respect to the cubic perovskite subcell (Fig. 4). It must be noticed that, when this type of pattern was observed, good images could not be obtained because the crystals were too thick.

A second type of boundary (APB) is often observed in the crystals, it corresponds to a relative translation of $b/2$ of the adjacent domains (Fig. 5a). Again, the junction between two domains (Fig. 5b) can easily be explained by the presence of octahedra involving a slight variation of local stoichiometry; this type of boundary, parallel to \tilde{c} , is often parallel to \tilde{b} or $\langle 110 \rangle$ but may also wander in the bulk.

The formation of such domains appears as a general feature due to the ordering of vacancies in the cubic perovskite matrix: ordered domains, with different orientations, have also been observed in $\text{CaMnO}_{2.8}$ (3–5).

Sr₂Mn₂O₅-Type Domains and “Pseudo-cubic” Perovskite Domains

Crystals in which domains exhibit a different crystallographic cell were also observed. Figure 6 shows an example of micrograph which was obtained by observation along c ; this type of crystal exhibits two domains: the first one (I) exhibits the $\text{Sr}_2\text{Mn}_2\text{O}_5$ structure ($a \approx a_p \sqrt{2}$ and $b \approx 2a_p \sqrt{2}$), the other (II) corresponds to a perovskite cell without any superstructure ($a \approx a_p$). The boundary between the two domains is approximatively planar and corresponds to the $(120)_p$ plane of the perovskite cell. It must also be noted that

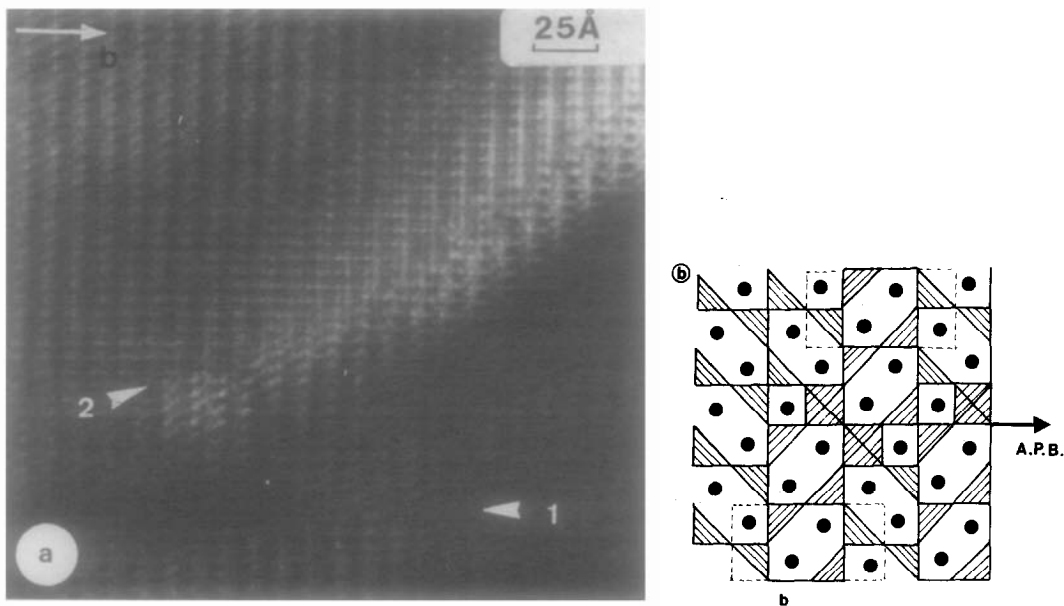


FIG. 5. Micrograph of $\text{Sr}_2\text{Mn}_2\text{O}_5$, along $[001]$, showing the existence of antiphase boundaries corresponding to a translation of $b/2$ of the adjacent domains (white arrows). (b) As an example, idealized drawing of an antiphase boundary, parallel to b with a translation of $b/2$.

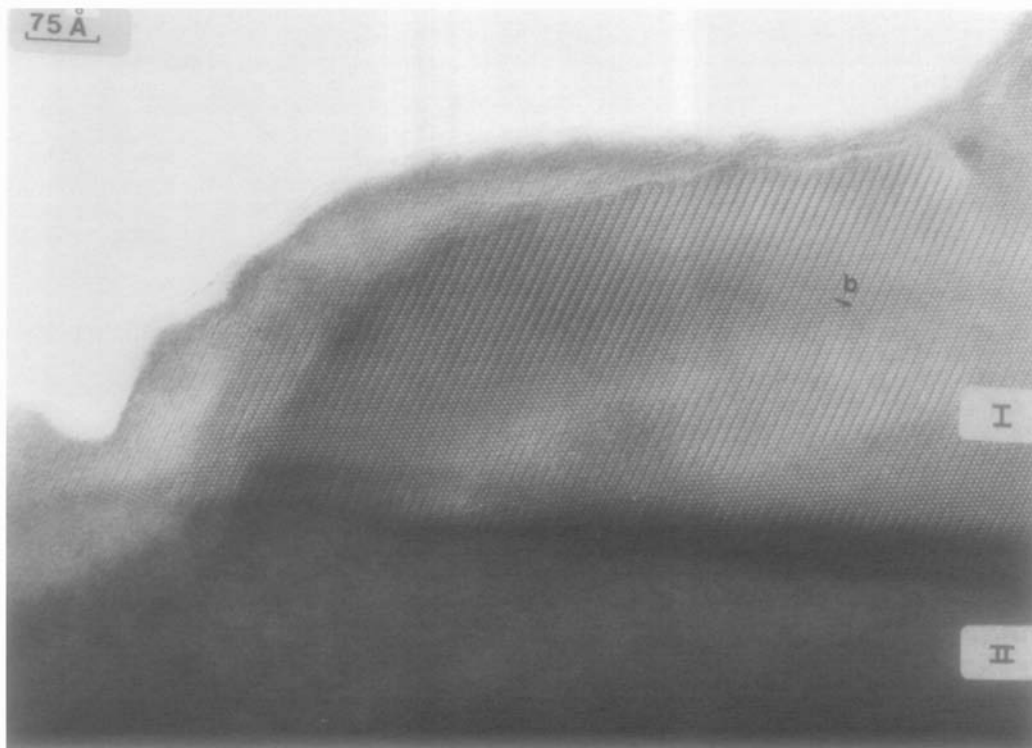


FIG. 6. Micrograph showing adjacent microdomains (projection along $[001]$): (I) with the $\text{Sr}_2\text{Mn}_2\text{O}_5$ -type structure and (II) without any superstructure. The corresponding electron diffraction pattern does not exhibit extra spots.

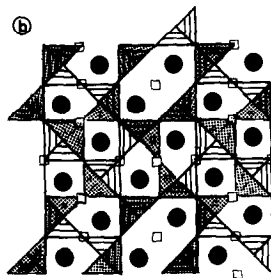
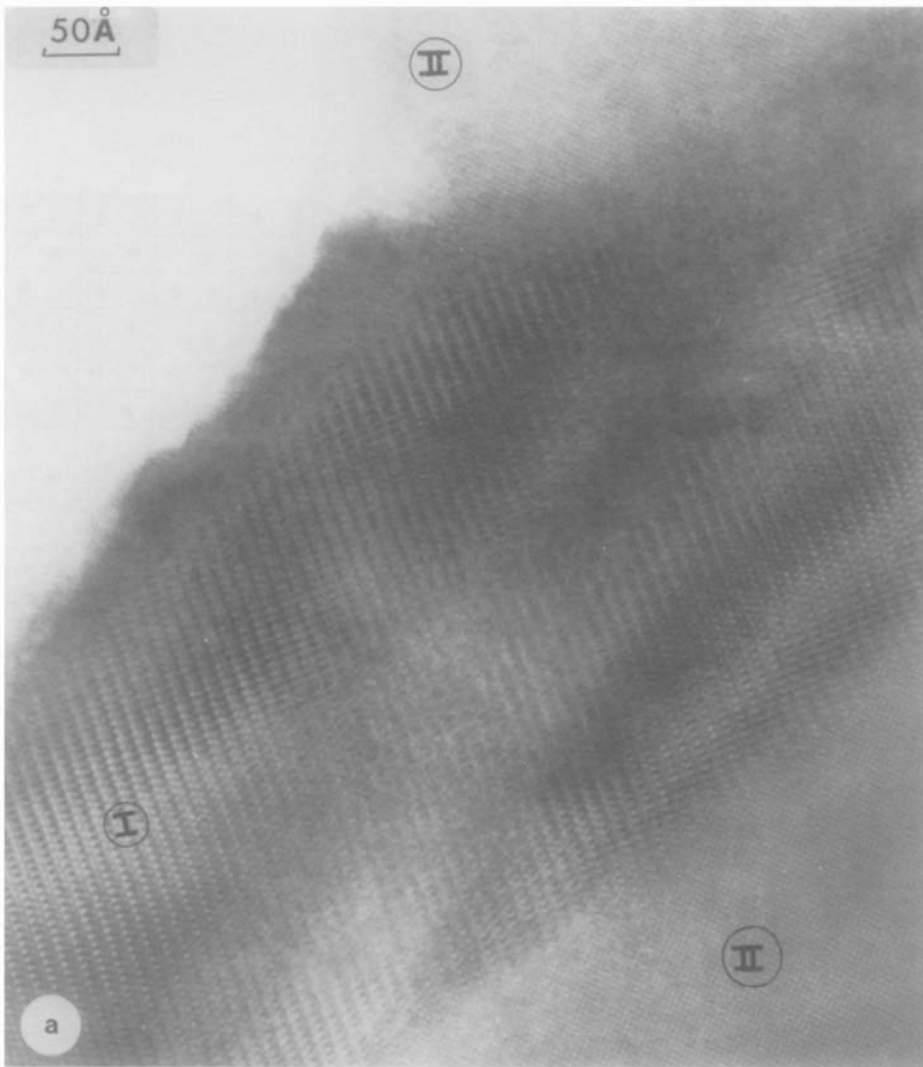


FIG. 7. Image of wandering boundaries between microdomains of type (I) and (II). (b) Idealized drawing of the superposition of two $\text{Sr}_2\text{Mn}_2\text{O}_5$ layers. The first layer corresponds to the hatched " Mn_2O_5 " pyramids, the second to the spotted ones. The angle between the two unit cells is 90° . Several orientations of two layers with respect to one another could thus be drawn.

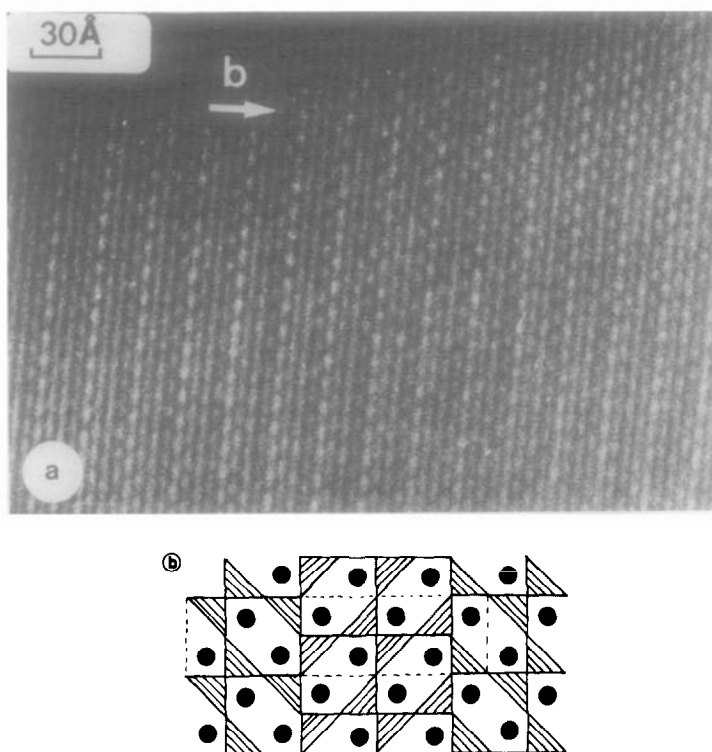


FIG. 8. Micrograph of a crystal showing regular fringes spaced $21.4 \text{ \AA} \approx 4a_p \sqrt{2}$. (b) Idealized model (5) for $\text{Ca}_2\text{Mn}_2\text{O}_5$ with $a \approx a_p \sqrt{2}$ and $b \approx 4a_p \sqrt{2}$.

the boundaries between the domains may wander as shown in Fig. 7a. It cannot be established whether the pseudocubic domains correspond to stoichiometric SrMnO_3 or to the composition $\text{Sr}_2\text{Mn}_2\text{O}_5$ or to intermediate compositions. However, taking into account the fact that the ordering of the oxygen vacancies can take place simultaneously in several directions (Figs. 3, 4) and that the global composition which was found by chemical analysis corresponds to $\text{Sr}_2\text{Mn}_2\text{O}_5$, suggests that the perovskite domains could also be built up of MnO_5 pyramids arranged in an aleatory manner, according to the composition $\text{Sr}_2\text{Mn}_2\text{O}_5$. For instance, the structure could be built up from a disordered stacking of $\text{Sr}_2\text{Mn}_2\text{O}_5$ layers along c , since two successive layers can be translated along different

directions with respect one to the other as shown, for example, in Fig. 7b.

Other Ordered Domains

Several crystals exhibit, besides the $\text{Sr}_2\text{Mn}_2\text{O}_5$ -type or cubic perovskite domains, ordered domains characterized by different superstructure cells. Such an example is shown on Fig. 8a; one observes a part of a crystal characterized by regular fringes spaced 21.4 \AA . This domain can easily be interpreted in terms of an other form of $\text{Sr}_2\text{Mn}_2\text{O}_5$ whose superstructure cell is characterized by the following parameters: $a \approx a_p \sqrt{2}$, $b \approx 4a_p \sqrt{2}$, and $c \approx a_p$. The structure of this form previously proposed for $\text{Ca}_2\text{Mn}_2\text{O}_5$ (5), shows that the composition of the crystal may remain unchanged, whereas the MnO_5 pyramids are connected

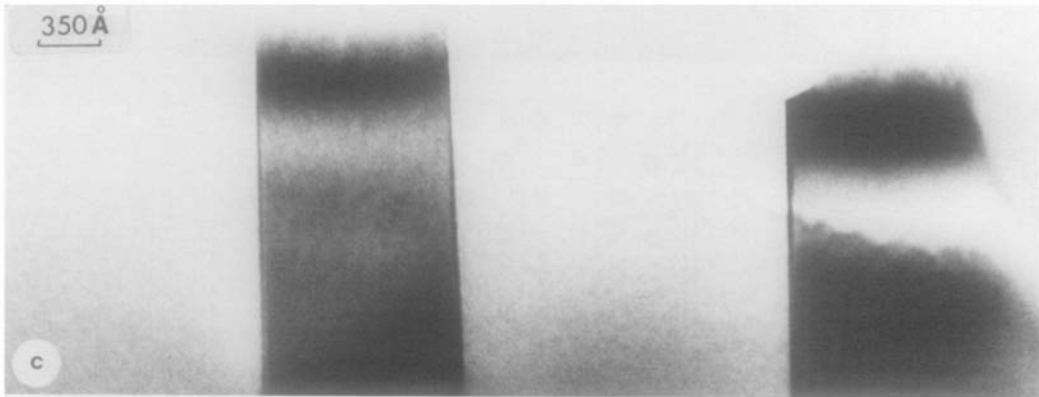
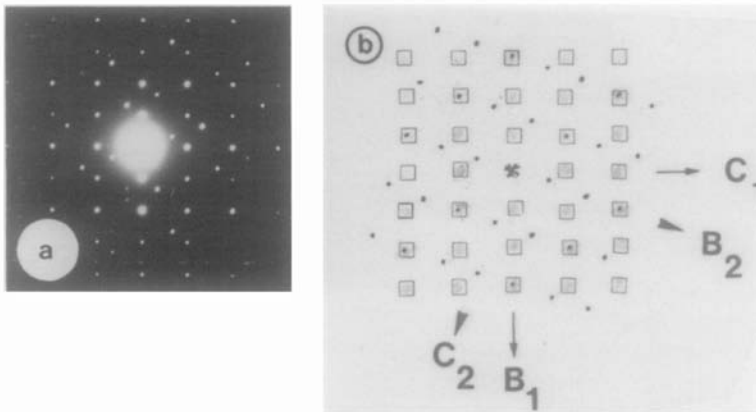


FIG. 9. Observed (a) and schematic (b) electron diffraction patterns of a twined crystal. The twine plane correspond to $(111)_p$ of the ideal cubic perovskite subcell. (c) Low-resolution image of the twined domains.

through their corners in a different manner (Fig. 8b).

Stacking Faults

The electron diffraction patterns of (100) show that some crystals are twined (Fig. 9a) as explained from the schematic drawing of the different zones (Fig. 9b). The corresponding low-resolution image (Fig. 9c) shows domains of various thicknesses ranging from 10 to 150 nm, and characterized by a twine plane parallel to $(111)_p$ with respect to the perovskite subcell. This type of defect can be interpreted as a stacking fault of the SrO_{3-x} layers, corresponding to the sequence $|\dots ccccc h ccccc \dots|$ as shown in

Fig. 10. This interpretation is in agreement with the existence of the hexagonal perovskites SrMnO_3 (6, 11, 12), whose

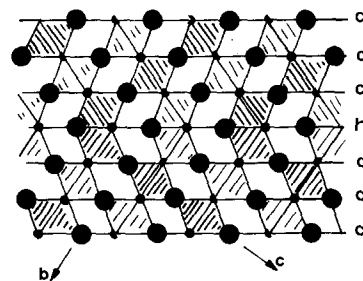


FIG. 10. Idealized drawing of a stacking fault of the SrO_{3-x} layers (large black circles are strontium atoms; small black circles, oxygens; and half circles correspond to $\frac{1}{2}\text{O}$, $\frac{1}{4}\square$ files of the MnO_5 pyramids).

SrO₃-layer stacking involves face-sharing MnO₆ octahedra. In the case of our crystals it cannot be decided whether these faults correspond to a local variation of oxygen content corresponding to the formation of SrO₃ layers or to a simple translation of ordered oxygen-defect SrO_{2.5} layers.

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